# University of Massachusetts Amherst

### Abstract

The purpose of this work is to identify an affordable and simple chromatographic technique to perform carbon monoxide chemisorption on dynamic supported metal catalysts for dispersion calculations.

### Background

### **Supported Metal Catalysts**

- industries Jsed variety 10 pharmaceuticals to the automotive industry
- Consist of metal catalyst nanoparticles loaded onto a support of a different material.

#### **Catalyst Dispersion**

- Ratio of exposed surface metal moles to the total amount of metal moles loaded on the support
- Only the surface of a catalyst can play a role in a reaction, thus greater dispersion creates greater catalytic reactions

### **Dynamic Chemisorption**

- Catalyst is exposed to a probe molecule gas in repeated pulses, CO in our experiment
- Pulses allow probe molecules to adsorb to unoccupied surface metal molecules
- Unadsorbed molecules are detected at the outlet

## We thank the iCons Sponsors:

# iCons: Dynamic Chromatographic Analysis of Metal Catalysts Nicholas Capra<sup>1</sup>, Shreya Thakkar<sup>1</sup>, Omar A. Abdelrahman<sup>1,2</sup> <sup>1</sup>Department of Chemical Engineering, University of Massachusetts Amherst

<sup>2</sup>Catalysis Center for Energy Innovation, University of Delaware

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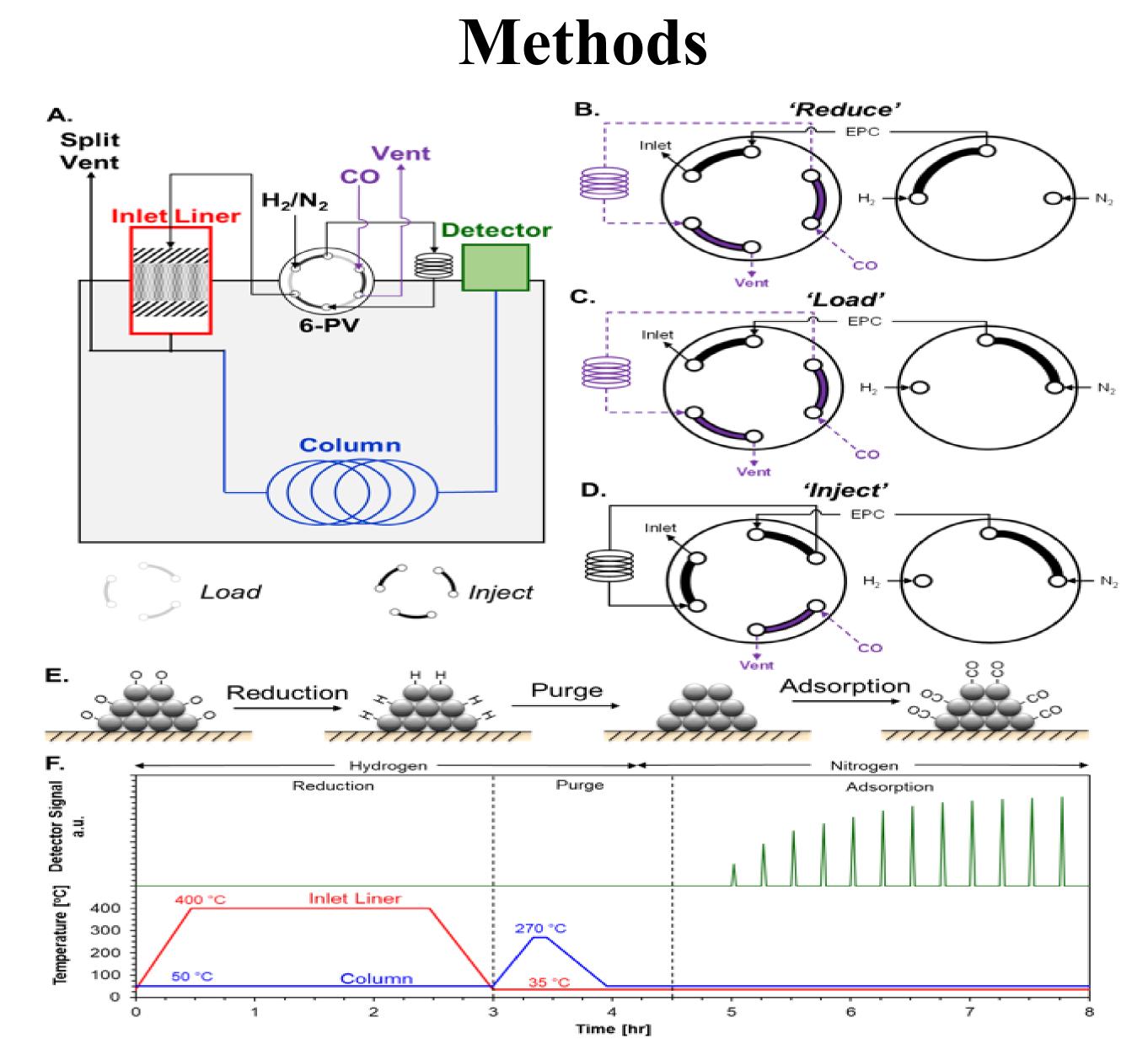


Figure 1: (A) Schematic of Gas Chromatograph with a 6-port valve for CO pulsing over the catalyst housed in inlet liner. (B) Valve position to perform reduction. (C) Load position, filling sample loop with CO sample. (D) Inject position, pulsing CO by pushing nitrogen gas through sample loop. (E-F) Three stages for Dispersion estimation of catalyst.

### **Technology Comparison**

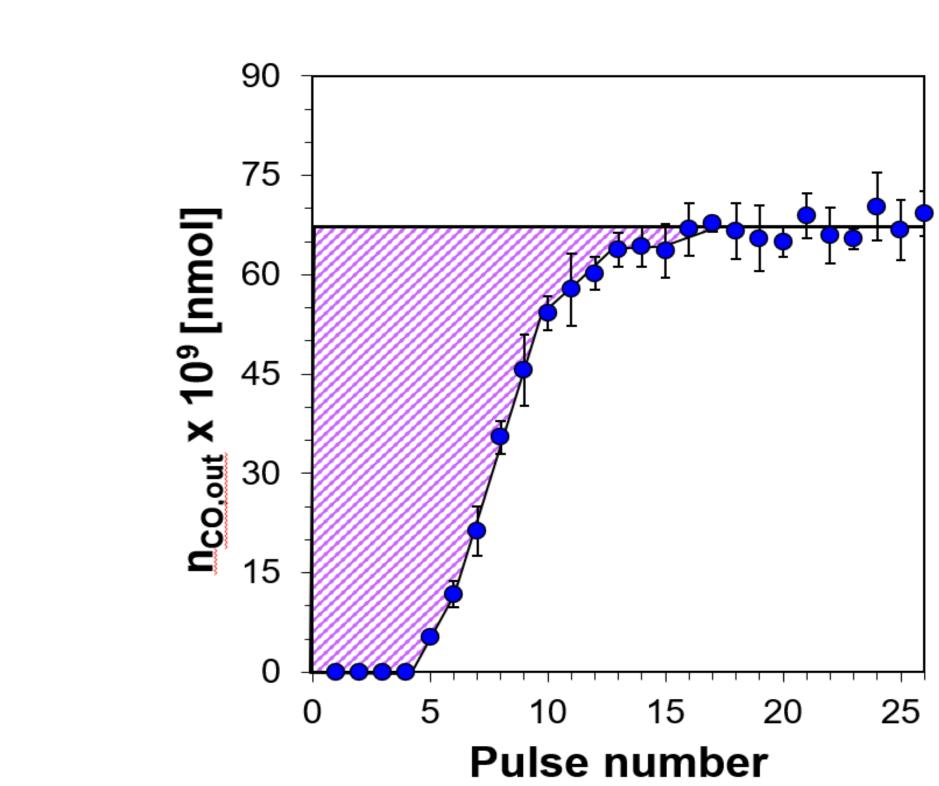
### Gas Chromatograph

- Cost: \$15K
- Requires additional port valves for multiple probe gases
- Can be used for other experiments Micromeritics AutoSorb
- Cost: \$70K
- Simultaneous sample preparation
- Perform physisorption as well as chemisorption

**Chleck Family Foundation** 

Edward Marram & Karen Carpenter





**Figure 2:** Breakthrough curve of 1wt% Platinum on Silica at 35° C

Catalyst	<b>Dispersion (%)</b>
1 wt% Pt/SiO <sub>2</sub>	27±2
$1 \text{ wt\% Pt/Al}_2\text{O}_3$	$32\pm3$
5 wt% Pt/C	$19 \pm 1$
0.12 wt% Pt/SiO <sub>2</sub>	$52\pm5$
Table 1: Dispersion measurements for supported Metal Catalysts	

- process
- treatments

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### Results

#### Automation

• Existing Chromatograph software automates the

• Port-valves are electronically actuated via signals sent to the apparatus to vary gases for pulsing and pre-

Integration of peaks can be automatically taken

#### Acknowledgements

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